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Kerala Plus Two Chemistry Chapter Wise Questions and Answers Chapter 11 Alcohols, Phenols and Ethers. Very Short Answer Type Questions (Score 1) Question 1. Glycerol is _____. Answer: 2, 3, 1 dihydroxy propanal. Question 2. Picric acid is a. 2, 4, 6 - trinitrotoluene

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Kerala Plus Two Chemistry Notes Chapter 3 Electrochemistry. Electrochemistry. It is the study of production of electricity from I energy which is released during spontaneous chemical reactions, also the use of electrical energy to bring about non-spontaneous chemical transformations. Electrolyte.

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Plus Two Chemistry Chapter Full Notes Kerala Plus Two Chemistry Notes Chapter 3 Electrochemistry. Electrochemistry It is the study of production of electricity from I energy which is released during spontaneous chemical reactions, also the use of electrical energy to bring about non-spontaneous chemical trasformations.

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$2 \text{Fe(s)} \rightarrow 2 \text{Fe}^{2+} + 4\overline{\text{e}}$ $E^\circ = -0.44 \text{ V}$. Electrons released at anodic spot move through metal and go to another spot on the metal and reduce O_2 in presence of H^+ . This spot behaves as cathode. $\text{O}_2(\text{g}) + 4\text{H}^+(\text{aq}) + 4\overline{\text{e}} \rightarrow 2 \text{H}_2\text{O(l)}$ $E^\circ = 1.23 \text{ V}$. The overall reaction is, $2 \text{Fe(s)} + \text{O}_2(\text{g}) + 4\text{H}^+(\text{aq}) \rightarrow 2 \text{Fe}^{2+} + 2\text{H}_2\text{O(l)}$ $E^\circ = 1.67\text{V}$

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Kerala Plus Two Chemistry Notes Chapter 2 Solutions. Solution A homogeneous mixture of two or more sub-stances is knows as solution. Solute : The substance present in smaller amount in a solution is called solute. Solvent: The substance present in larger amount in a solution is called solvent.

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Kerala Plus Two Chemistry Notes Chapter 9 Coordination Compounds. Coordination chemistry \rightarrow branch of chemistry which deal with the complex compounds formed by tmasition and other metals. Chlorophyll, haemoglobin and vitamin B 12 are coordination compounds of Mg, Fe and Co respectively. Werner's Theory of Coordination Compounds The main postulates are,

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$2\text{Ca(OH)}_2 + 2\text{Cl}_2 \rightarrow \text{Ca(OCl)}_2 + \text{CaCl}_2 + 2\text{H}_2\text{O}$ $\text{Cl}_2 + \text{H}_2\text{O} \rightarrow 2\text{HCl} + [\text{O}]$ Coloured substance + $[\text{O}] \rightarrow$ colourless substance. Uses: For bleaching wood pulp, cotton and textiles; for the preparation of insecticide, pesticides and other organic solvents, e.g. CHCl_3 , DDT, BHC etc. Hydrogen Chloride (HCl): Preparation:

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