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atom's naturally occurring isotopes. 2.2 Silicon has three ...

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metal, the equilibrium
number of vacancies at 900 C
is $2.3 \times 10^{25} \text{ m}^{-3}$. If the
density and atomic weight of
this metal are 7.40 g/cm^3
and 85.5 g/mol ,
respectively, calculate the
fraction of vacancies for

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this metal at 900 C.

Solution This problem is solved using two steps: (1) calculate the total number of lattice sites in silver, N , using

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