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Substituting and multiplying by the factor 109 for the density unity  $\text{kg}/\text{km}^3$ , the mass of the atmosphere is determined to be  $m = 5.092 \times 10^{18} \text{ kg}$  Discussion Performing the analysis with excel would yield exactly the same results. EES Solution for final result:  $a = 1.2025166$   $b = -0.10167$   $c = 0.0022375$   $r = 637$

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$7h=25m=4\pi(a r^2 h+r(2a+br)h^2/2+(a+2br+cr^2)h^3/3+(b+2cr)h^4/4+ch^5/5)*1E+9$  1-7 Pressure, Manometer, and Barometer 1-34C The pressure relative to the atmospheric pressure is called ...

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Solution The volume and the weight of a fluid are given. Its mass and density are to be determined. Analysis Knowing the weight, the mass and the density of the fluid are determined

to be  $32225 \text{ N}$ ,  $1 \text{ kg}$ ,  $m/s$ ,  $9.80 \text{ m/s}$ ,  $1 \text{ N}$ ,  $W$ ,  $m$ ,  $g$

$== =$   $23.0 \text{ kg}$ ,  $23.0 \text{ kg}$ ,  $24 \text{ L}$ ,  $m$   $==$

$=0.957 \text{ kg/L}$ ,  $V$

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based file parameters. Analysis The problem is solved using EES, and the solution is given below. Properties The heat of fusion of water at atmospheric pressure is.

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Assumptions. At specified conditions, air behaves as an ideal gas.

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The results are:  $(z) = a + bz + cz^2 = 1.20252 - 0.101674z + 0.0022375z^2$  for the unit of kg/m<sup>3</sup>, (or,  $(z) = (1.20252$

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–  $0.101674z + 0.0022375z^2$ )  $\times 10^9$  for the unit of  $\text{kg}/\text{km}^3$ )  
where  $z$  is the vertical distance from the earth surface at sea  
level. At  $z = 7$  km, the equation gives  $\rho = 0.600 \text{ kg}/\text{m}^3$ .

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